

## Chapter 6

### Conclusions and Recommendations

#### 6.1 Conclusions

1. The industry has already implemented a few waste minimization activities such as quality control of intermediate product (polished rims) with in the process, two stage rinsing, dragout recovery.
2. However, the industry lacks technical know-how to carry out the electroplating process efficiently. Quality control of the product is poor.
3. There are possibilities to reduce electricity and water consumption without any quality deterioration of the product; through working longer hours a day, introduction of temperature control units for heaters and removal of unnecessary unit operations. The study showed that a saving of at about Rs.2.40 per rim could be achieved by increasing operating time by 4 hrs (i.e. 12hrs operation per day).
4. Wastewater generated in the surface preparation process is unnecessarily containing nickel and chromium, due to improper preparation of rejects before recycling into the process, which can be eliminated by introduction of stripping for rejects before recycling. The study showed that a saving of at about Rs.0.5 per rim could be achieved by this modification.



5. In the wastewater treatment, neutralization should be carried out only as the final step; neutralization before sedimentation of the precipitated nickel compounds re-dissolves them, which results in higher nickel concentrations in the effluent.
6. Optimum pH for hydroxide precipitation of nickel and chromium in a mixture is pH 8.


## 6.2 Recommendations

1. The following waste minimization activities can be implemented immediately.

- Introduction of stripping for rejects
- Introduction of drain boards
- Introduction of temperature feed back control system for heaters
- Introduction of improved process control in Ni plating activity
- Increasing working hours a day.

2. Following waste minimization activities may be implemented, while proper quality control procedures such as corrosion resistant test & Adhesion tests are carried out.

- Omission of unnecessary unit operations
- Change over to  $\text{Cr}^{3+}$  chrome plating solution instead of  $\text{Cr}^{6+}$  solutions

3. As recommended in the literature unanimously, it is recommended to carryout chromium reduction at pH 3 for minimum of 30 minutes time period after adding the required amounts of H<sub>2</sub>SO<sub>4</sub> acid and Sodium Meta Bisulphite.
  4. It is recommended to test a sample of reduced chromium waste using KI / starch paper or KI/starch solution before releasing that into Heavy Metal Precipitation Tank in order to check whether any Cr<sup>6+</sup> is remaining in that. If the test is positive for Cr<sup>6+</sup> recommendation 3 should be repeated.
  5. It is recommended to carry out nickel and chromium precipitation using NaOH or Ca(OH)<sub>2</sub> at a pH value between 8 – 8.5.
- 

University of Moratuwa, Sri Lanka.  
Electronic Theses & Dissertations  
www.lib.mrt.ac.lk
6. Sedimentation should be carried out before neutralization of the wastewater
  7. Sedimentation may be improved using Ca(OH)<sub>2</sub>,or Alum while maintaining the optimum pH for precipitation of the heavy metals present in the wastewater.
  8. It is recommended to filter Supernatant of the sedimentation tank since even a small amount of suspended material that passes with the supernatant can lead to exceeded heavy metal concentrations in the treated wastewater beyond the allowed discharge standards.

## REFERENCES

ANONYMOUS. (1991), Natural Resources of Sri Lanka- Conditions and Trends. A report prepared for the Natural Resources, Science and Energy Authority of Sri Lanka. PP 139.

AREWGODA, M.C. (1986), *Pollution due to Heavy Metal in the Kelaniya Area*. in Dissanayaka, C. B. and Gunatilaka, L. (Eds): *Some Aspect of the Chemistry of the Environment of Sri Lanka*. pp.161-164, Sri Lanka Association for the Advancement of Science (Colombo)

ARUN KUMAR, L.C., VINCENT, S., AMBROS, T., (1994), Uptake and Persistence of the Heavy Metal Cadmium in Tissues of the Fresh Water Fish *Cyprinus Carpio* Poll. Res.13(4) : 361-364

BATSTONE, R., SMITH, J.E., and WILSON, D. (1989), *The Safe Disposal of Hazardous Wastes. The special Needs and Problems of Developing Country. Volume I, World Bank Technical Paper Number 93*. pp.31-32, 161-184, 270-440 The International Bank for Reconstruction and Development the World Bank (Washington D.C)

BMU, (1992), *A-Z, of Environmental Protection in Germany. A Pocket Book for Technical and Supervisory Staff in Industry, Commerce Education the Media and Administration*. pp.76, Environmental Information Center (Dusseldorf)

BRIMI, M.A. and LUCK, L.R. (1965) *Electrofinishing*. American Elsevier Publishing Co. Inc.(New York)

CEA, (1992), *Industrial Pollution Control Guidelines, No.8, Metal Finishing Industry*, pp.1-21, Central Environmental Authority. Ministry of Environmental and Parliamentary Affairs, Sri Lanka (Colombo)

CRITENDEN, B. and KOLACZKOWSKI, S. (1995), *Waste Minimization A Practical Guide*. Institution of Chemical Engineers, Rugby ,U.K..

DALLINGER, R., PROSSI, F., SEGNER, H. and BACK, H. (1987), Contaminated Food and Uptake of Heavy metals in Fish: A review and proposal for further Research. *Oecologia* 73 P91-98.

DUFFY, J.I., (1981), *Electroplating Technology – Recent Developments* pp181-188, Noyes Data Corporation Park Ridge New Jersey USA.

EPA, (1980), *An exposure and risk assessment for zinc* pp.180, Environmental Protection Agency

FISH, R.A. (1977), *Toxic and Other Hazardous Waste* Publication No ICP/CEP 402 (CVIII), World Health Organization Regional Office for Europe (Copenhagen).

GREENBERG, A.E., CLESCEN, L.S. and EATER, A.D. (1992), *Standard Methods for Examination of Water and Waste Water 18th Edition* pp.3-1 to 3-18, American Public Health Association and American Water Works Association (Washington)



University of Moratuwa, Sri Lanka.  
Electronic Theses & Dissertations  
[www.lib.mrt.ac.lk](http://www.lib.mrt.ac.lk)

HIRST, J. M. and ASTON, A.S.R. (1983), Behaviour of Copper, Zinc, Iron and Manganese during Experimental Resuspension and Reoxidation of polluted Anoxic Sediments. *Estuarine, Coastal and Shelf Science*, 16, 549 - 558

JICA (1997), *Study on Industrial Sector Development (Electroplating and Waste Water Treatment in the Democratic Socialist Republic of Sri Lanka)*, Tokyo, Japan.

JORGENSON S.E., (1979), *Industrial Wastewater Management*, Elsevier Publishing Co. Inc (New York) USA.

KUHN, A.(19??) *Industrial Electrochemical Processes*, American Elsevier Publishing Co. Inc (New York)

LECKIE, J.O., and JAMES, R.O.(1974) Control Mechanisms for Trace Metals in Natural Waters. In A. J. RUBINS (Ed.) *Aqueous Environmental Chemistry of Metals*. Ann Arbor Science Michigan p.1.

METELEV, V.V., KANADEV, A.I. and DZASOKKOVA, N.G. (1983), *Water Toxicology* P 86, Amerind Publishing Co. Pvt. Ltd., (New Delhi)

MENDIS, S. (1986), *The significance of Trace Elements in Human Health* in Dissanayaka, C.B. and Gunatilaka, L. (Editors): *Some Aspect of the Chemistry of the Environment of Sri Lanka* pp.25-138, Sri Lanka Association for the Advancement of Science (Colombo)

MEPA, (1991), *The United Nations Conference on Environmental and Development*. Sri Lanka National Report, Ministry of Environment & Parliamentary Affairs, Sri Lanka (Colombo)

MILLER, A., (1970), Carcinogenesis by chemicals: An overview- G.H.A. Clawes Memorial Lecture, *Cancer Research.*, 30, P559-579.



University of Moratuwa, Sri Lanka.  
Electronic Theses & Dissertations  
[www.lib.mrt.ac.lk](http://www.lib.mrt.ac.lk)

PASSOW, H., ROMSTEIN, A., and CLARKSON, T.W. (1961). *The General Physiology of Heavy Metals*. *Pharmacology Rev.* 13 pp 185-225

PLETCHER DEREK (1982), *Industrial Electrochemistry* pp. 172-199, Chapman and Hall Ltd., London

PONNAMBALAM, M. (1986), *Chromium in the Environment - survey* pp.65-78, in Dissanayaka, C.B. and Gunatilaka, L. (Editors): *Some Aspect of the Chemistry of the Environment of Sri Lanka*. Sri Lanka Association for the Advancement of Science (Colombo)

RUNE LONNGREN (1992), *Institutional approaches to chemicals control - A historical overview* P99, 166, The National Chemicals Inspectorate, Sweden.

STEWART, E. (1989), *Chemical Analysis of Ecological Materials* pp.31, Second Edition Blackwell Scientific Publications (Oxford)

SUSSES, M.J., and HUISMANS, J.W., (Eds):(1983), *Management of Hazardous Waste, Policy Guidelines And Code of Practice*, WHO office for Europe, Copenhagen.

SUNIL, P.P. (1976), *The Heavy metals Science Report* pp.279-283, National Environmental Engineering Research Institute, India (Nagpur)

TONDON, O.N., (1984) *Hand Book of Electroplating, Anodizing and Metal Treatments*, Small Industry Research Institute, Delhi.



University of Moratuwa, Sri Lanka.  
Electronic Theses & Dissertations  
[www.lib.mrt.ac.lk](http://www.lib.mrt.ac.lk)

UN, (1987), *State of the Environment in Asia and Pacific Volume II* pp.274-275, United Nations, Economic and Social Commission for Asia and the Pacific Bangkok, (Thailand)

UNEP (1991) *Audit and Reduction Manual for Industrial Emission and Waste*, Technical Report Series No7, Industry and Environment office, France.

WHO, (1996)., *Guidelines for Drinking Water Quality (Vol.I) Recommendations*, pp.213,312 World Health Organization (Geneva)

WHO, (1988), *Chromium - Environment Aspects, Environmental Health Criteria 61* pp.49-69, IPCS International Programme on Chemical Safety. World Health Organization (Geneva)

WHO, (1991)., *Nickel - Environmental Aspects, Environmental Health Criteria 108* pp.49-79, IPCS

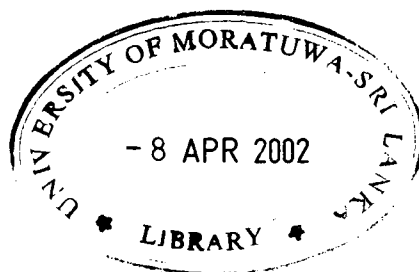
International Programme on Chemical Safety, World Health Organization (Geneva)

WHO, (1996), *Fundamentals of Applied Toxicology, The Nature of Chemical Hazardous*, pp.81-88, International programme on Chemical Safety, World Health Organization (Geneva)

ZABEL, T.F. (1993)., Diffuse sources of pollution by Heavy Metals. *J. IWEM.*, 7, 513-520



University of Moratuwa, Sri Lanka.  
Electronic Theses & Dissertations  
[www.lib.mrt.ac.lk](http://www.lib.mrt.ac.lk)





CENTRAL ENVIRONMENTAL AUTHORITY

GENERAL STANDARDS FOR DISCHARGE OF  
EFFLUENTS INTO INLAND SURFACE WATERS

No.	Determinant	Tolerance limit
1	Total suspended solids, mg/l max	50
2	Particle size of total suspended solids	Shall pass sieve of aperture size 850 micro m.
3	pH value at ambient temperature	6.0 to 8.5
4	Biochemical Oxygen Demand — BOD <sub>5</sub> in five days at 20°C, mg/l, max	30
5	Temperature of discharge	Shall not exceed 40°C in any section of the Stream within 15 m down stream from the effluent outlet
6	Oils and grease, mg/l max	10.0
7	Phenolic Compounds (as phenolic OH) mg/l, max	1.0
8	Cyanides as (CN) mg/l, max	0.2
9	Sulfides, mg/l max	2.0
10	Fluorides, mg/l max	2.0
11	Total residual chlorine mg/l max	1.0
12	Arsenic, mg/l, max	0.2
13	Cadimium total mg/l, max	0.1
14	Chromium total, mg/l, max	0.1
15	Copper total, mg/l, max	3.0
16	Lead, total, mg/l, max	0.1
17	Mercury total, mg/l, max	0.0005
18	Nickel total, mg/l, max	3.0
19	Selenium total, mg/l, max	0.05
20	Zinc total, mg/l, max	5.0
21	Ammoniacal nitrogen, mg/l, max	50.0
22	Pesticides	Undetectable
23	Radio active material	
	a Alpha emitters $\mu\text{C/ml}$	10 <sup>-7</sup>
	b Beta emitters $\mu\text{C/ml}$	10 <sup>-8</sup>
24	Chemical Oxygen Demand (COD), mg/l, max	250

Note 1 All efforts should be made to remove colour and unpleasant odour as far as practicable.

Note 2 These values are based on dilution of effluents by at least eight volumes of clean receiving water. If the dilution is below eight times, the permissible limits are multiplied by one-eighth of the actual dilution.

Note 3 The above mentioned General Standards shall cease to apply with regard to a particular industry when industry specific standards are notified for that industry.